A SHORT SYNTHESIS OF 4-(3-FURYL)-4-OXOBUTANOIC ACID AND SYNTHESIS OF EGOMAKETONE

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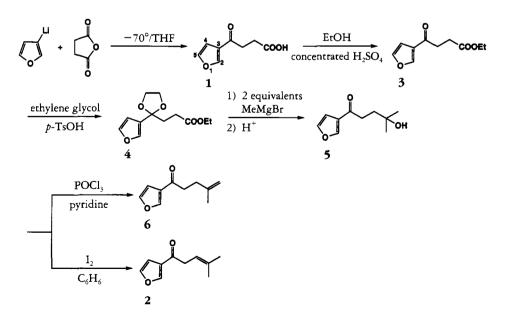
ABSTRACT.—3-Lithiofuran in THF was added to succinic anhydride in THF to yield 4-(3furyl)-4-oxobutanoic acid [1]. Egomaketone [2], isolated from *Perilla frutescens* was synthesized from 1 in four steps. The isomer of 2, 5-(3-furyl)-2-methyl-5-oxo-1-pentene [6], was also prepared from 1.

Oxygenated perillens (perilla ketone, egomaketone, and isoegomaketone), isolated from Perilla frutescens Birt. (1-3), are a major cause of bovine respiratory toxicity (4). The syntheses of compounds belonging to this group of furan monoterpenoids have been accomplished by various methods. Egomaketone [2] has been prepared via 3-cyanofuran (5, 6), 3-furoyl chloride(7), 3-furancarbaldehyde (8), 2-(3-furyl)-1,3-dithiane (9), and 3furyltrimethyltin (10), using 3bromofuran as starting material. However, some of these syntheses involve many steps and others use expensive reagents to introduce a side-chain onto the 3-position of furan.

The preparation of 2-(3-furanoyl)-

benzoic acid from 3-lithiofuran has been effected by our group in only one step (11). In the present paper, we report a one-step synthesis of 4-(3-furyl)-4oxobutanoic acid [1], an intermediate in the preparation of oxygenated perillens, from 3-lithiofuran. The previously reported synthesis of 1 requires five steps from diethyl 3,4-furandicarboxylate(12). Further, we report a new synthetic route from 1 to egomaketone [2] (Scheme 1).

The intermediate **1** was obtained by the reaction of succinic anhydride with 3lithiofuran in one step. Acid **1** was esterified to give ethyl 4-(3-furyl)-4oxobutanoate [**3**], and the ketone was protected as the ketal, ethyl 4-(3-furyl)-4-(1,3-dioxolan-2-yl)butanoate [**4**]. Re-



SCHEME 1. Syntheses of 5-(3-furyl)-2-methyl-5-oxo-1-pentene [6] and egomaketone [2] from 3-lithiofuran.

action of 4 with two equivalents of methylmagnesium bromide, followed by treatment with iced 10% HCl, yielded the tertiary alcohol, 5-(3-furyl)-2-methyl-5-oxo-2-pentanol [5]. The heat-sensitive alcohol 5 was used for the next reaction without further purification.

The alcohol **5** was allowed to react with phosphorus oxychloride (13) in pyridine. The reaction mixture was analyzed by gc-ms which detected the starting material and two dehydrated compounds. The mixture was purified by cc to afford white needles of 5-(3-furyl)-2methyl-5-oxo-1-pentene [**6**] (mp 34–36°) in poor yield. The ¹H-nmr spectrum of **6** showed characteristic signals for the exomethylene group at δ 4.70 (s) and a methyl group at δ 1.77 (s).

The alcohol **5** was then reacted with iodine in C_6H_6 (14). The reaction mixture was analyzed by gc-ms, which detected the starting material and two dehydrated compounds. The mixture was purified by cc to give egomaketone [**2**] in low yield; spectral data for **2** were in accord with those of an authentic sample (6).

In conclusion, the key intermediate 1 was obtained from 3-lithiofuran in one step. Egomaketone 2, isolated from *Perilla frutescens* Birt., and the isomer 6 were prepared from 1. It has been demonstrated that this method affords a new route to 3-acylfurans.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—All mps (open capillaries) were determined using a Yamato MP-21 apparatus and are reported uncorrected. All boiling points are also reported uncorrected. The ¹H-nmr spectra were determined at 60 MHz on a Nippon Denshi JNM PMR-60SI nmr spectrometer with TMS as internal reference. The ir spectra were measured using a Jasco IR-810 spectrometer and the mass spectra were obtained on a Nippon Denshi DX-300 spectrometer at 70 eV. THF was distilled over LiAlH₄ prior to use. Chromatography was carried out using Si gel (Wakogel C-200, Wako Pure Chemical Industries, Ltd.). Methylmagnesium bromide (ca. 3 M in Et₂O, Tokyo Chemical Industry Co., Ltd.) and *n*-butyllithium (1.68 M in hexane, Kanto Chemical Co., Inc.) were obtained commercially.

4-(3-Furyl)-4-oxobutanoic acid [1].-3-Bromofuran (21.6 g, 147 mmol) in THF (180 ml) was added to n-butyllithium (110 ml of 1.68 M in hexane, 185 mmol) with stirring at -70° under an atmosphere of N2. The mixture was allowed to react for 10 min. The mixture was then added to succinic anhydride (16.2 g, 162 mmol) in THF (240 ml) at -70° under an atmosphere of N₂ for 1 h. The mixture was warmed to room temperature for 30 min and poured into ice-cold H₂O. The solution was made acidic with 10% HCl and extracted with Et₂O. The desired compound 1 was extracted with 5% NaHCO₃ from the Et₂O layer. The aqueous solution was made acidic with 10% HCl and extracted with Et₂O. The Et₂O layer was washed with brine and dried over anhydrous Na₂SO₄. The solution was evaporated, and the residue was recrystallized from H₂O to yield 1 as white scales (11.7 g, 70 mmol, 47% yield): mp 146–147° [mp 148–150° (15)]; ir ν max (KBr) 1720 (COOH), 1660 (C=O) cm^{-1} ; ¹H nmr (DMSO- d_6) δ 12.17 (1H, s, OH, exchangeable proton), 8.53 (1H, d), 7.73 (1H, t), 6.75 (1H, d), 3.05 (2H, t, J=7 Hz, COCH₂), 2.55 (2H, t, J=7 Hz, CH₂COOH); ms m/z [M]⁺ 168 (10), 150 (14), 95 (100); anal., calcd for C8H8O4, C, 57.14, H, 4.80; found C, 57.10, H, 4.74.

Ethyl 4-(3-furyl)-4-oxobutanoate [3].-To compound 1 (10.0 g, 60 mmol) in absolute EtOH (35 ml), concentrated H₂SO₄ (0.7 ml) was added slowly, and the mixture was refluxed for 4 h. Excess EtOH was evaporated, and the resulting mixture was poured into ice-cold H₂O. The solution was extracted with Et₂O. The Et₂O layer was washed with 5% NaHCO₃, then brine, and dried over anhydrous Na2SO4. The solvent was evaporated, and the residue was recrystallized from EtOH/H₂O to yield **3** as white needles (9.0 g, 46 mmol, 77% yield): mp 53-55°; ir v max (KBr) 1720 (COOEt), 1670 (C=O) cm^{-1} ; ¹H nmr (CDCl₃) δ 8.03 (1H, d), 7.73 (1H, t), 6.73 (1H, d), $4.11 (2H, q, J=7 Hz, CH_2 CH_3), 3.07 (2H, t, J=7$ Hz, $COCH_2$), 2.72 (2H, t, J=7 Hz, CH_2COOEt), 1.23 (3H, t, J=7 Hz, CH₂CH₃), ms m/z [M]⁺ 196 (5.1), 151 (15), 95 (100); anal., calcd for C₁₀H₁₂O₄, C, 61.22, H, 6.16; found, C, 61.06, H, 6.16.

Ethyl 4-(3-furyl)-4-(1,3-dioxolan-2-yl)butanoate [4].—Ethylene glycol (18.0 ml) and p-toluenesulfonic acid (0.4 g) were added to compound 3 (10.0 g, 51 mmol) in ethyl orthoformate (34.0 ml), then the mixture was heated at 165° for 12 h. The mixture was cooled to room temperature and diluted with Et₂O. The Et₂O layer was washed with brine and dried over anhydrous Na₂SO₄, then the solvent was evaporated. The residue was distilled to yield 4 as a colorless liquid (10.5 g, 44

mmol, 86% yield): bp 126° (3 mm Hg); ir ν max (neat) 1735 (COOEt) cm⁻¹; ¹H nmr (CDCl₃) δ 7.30 (2H, d), 6.30 (1H, d), 4.12 (2H, q, J=7 Hz, CH₂CH₃), 3.90 (4H, m, dioxolane), 2.32 (4H, m, CH₂CH₂), 1.21 (3H, t, J=7 Hz, CH₂CH₃); ms m/z[M]⁺ 240 (0.4), 195 (15), 139 (100); anal., calcd for C₁₂H₁₆O₅, C, 59.99, H, 6.71; found, C, 59.71, H, 6.61.

5-(3-Furyl)-2-methyl-5-oxo-2-pentanol [5].— Methylmagnesium bromide (6.1 ml of 3 M in Et₂O, 18.3 mmol) was diluted with absolute Et₂O (2 ml). To the solution, compound 4 (2.0 g, 8.3 l)mmol) in absolute Et₂O (6 ml) was added below 5° under an atmosphere of N2. The mixture was stirred at 0° for 2 h, then at room temperature for 2 h. The resulting mixture was poured into iced 10% HCl. The solution was extracted with Et₂O. The Et₂O layer was washed with brine and dried over anhydrous Na₂SO₄; then the solvent was evaporated to yield 5 as a colorless liquid (1.4 g, 7.7 mmol, 93% yield). Because alcohol 5 was sensitive to heat, it was used without further purification: ir ν max 3440 (OH), 1675 (C=O) cm⁻¹; ¹H nmr (CDCl₃) δ 8.10 (1H, d), 7.42 (1H, t), 6.73 (1H, d), 2.90 (2H, t, *J*=7 Hz, COCH₂), 2.83 (1H, s, OH, exchangeable proton), 1.85 (2H, $t, J=7 \text{ Hz}, CH_2C(Me)_2OH), 1.22 (6H, s, Me \times 2);$ hrms m/z calcd for $C_{10}H_{14}O_3$ 182.0943 ([M]⁺, 5.1), observed 182.0930, 167 (26), 124 (64), 95 (100).

5-(3-Furyl)-2-methyl-5-oxo-1-pentene [6].— To compound 5 (1.4 g, 7.7 mmol) in absolute pyridine (66 ml), POCl₃ (6.6 ml, 72 mmol) in absolute pyridine (20 ml) was added at -15- -20° ; the mixture was stirred at this temperature for 30 min and then at room temperature for 24 h. The mixture was poured into ice-cold H₂O and the solution was extracted with Et₂O. The Et₂O layer was washed with 10% HCl, H₂O, and brine. The Et₂O layer was dried over anhydrous Na₂SO₄, then the solvent was evaporated. The residue was purified by cc over Si gel with hexane-EtOAc (9:1) as eluent to yield 6 as white needles (0.1 g, 0.6 mmol, 8% yield, $R_f 0.34$): mp 34–36°; ir ν max (KBr) $1665 (C=O) \text{ cm}^{-1}$; ¹H nmr (CDCl₃) δ 8.02 (1H, d), 7.40 (1H, t), 6.73 (1H, d), 4.70 (2H, s, $CH_2 =$), 2.90 (2H, t, J=7 Hz, COCH₂) 2.38 (2H, t, J=7 $H_z, CH_2C(Me) = CH_2$, 1.77 (3H, s, Me); hrms m/zcalcd for $C_{10}H_{12}O_2$ 164.0837 ([M]⁺ 8.2), observed 164.0860, 95 (100).

Egomaketone [2].-To compound 5 (1.1 g,

6.0 mmol) in C₆H₆ (20 ml), iodine (0.2 g) was added and the mixture was refluxed for 1 h. The resulting mixture was poured into 10% Na₂S₂O₃. The solution was extracted with Et₂O. The Et₂O layer was washed with 10% Na₂S₂O₃ and brine. The Et₂O layer was dried over anhydrous Na₂SO₄, and the solvent was evaporated. The residue was purified by cc over Si gel with hexane-EtOAc (19:1) as eluent to yield 2 as a colorless liquid (0.2 g, 1.2 mmol, 20% yield, R, 0.18); ir v max 1675 $(C=O) \text{ cm}^{-1}$; ¹H nmr $(CDCl_3) \delta 8.05 (1H, s), 7.41$ (1H, t), 6.75 (1H, d), 5.38 (1H, dt, J=2 and 7 Hz,CH=), 3.45 (2H, d, J=7 Hz, COCH₂), 1.77 (3H, s, Me), 1.68 (3H, s, Me); hrms m/z calcd for $C_{10}H_{12}O_2 164.0837([M]^+, 12)$, observed 164.0858, 95 (100).

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